MATERIAL SAFETY DATA SHEET FOR OZONE GAS

Product Information				
Product Name		Ozone		
Synonyms		Triatomic Oxygen, O3		
Chemical Formula		03		
Description		Gaseous oxidant		
Molecular Weight		48.0		
Other Designations		None		
Hazardous Components				
Components		Ozone Gas		
Concentration		0-15% by weight		
CAS#		10028-15-6		
ICSC#		0068		
Physical Data				
Boiling Point (760mm Hg)		-111.9° C		
Melting Point		-192.7°C		
Gas Density (0°C and 1 atm.)		2.14 g/l		
Vapor Density (air=1)		1.6		
Water Solubility (20°C, 4% ozone in oxygen)		19 mg/l		
Specific Gravity		1.614		
Odor		Pungent		
Appearance and Odor:		Colorless gas with pungent odor generally detectable at 0.01 to 0.04ppm and a sharp disagreeable odor at 1.00ppm.		
Fire and Explosion Hazard				
Flash Point	Not Applicable			
Auto-Ignition Temperature	Not Applicable			
Flammability	Non-Flammable but enhances combustion of other substances. Some reactions may cause fire or explosion.			
Extinguishing Media	Use extinguishing media appropriate for the fuel source.			
Special Fire Fighting Procedures	Use self-contained breathing apparatus. Ozone is an oxidizer.			
Unusual Fire and Explosion Hazards	Ozone can react explosively with readily oxidizable substances and reducing agents.			

Reactivity Data					
Unstable Decomposes to form overgon under ordinant conditions thus is not encountered					
Stability	except in the immediate area where it is formed.				
Reactivity	Reacts with any materials that can oxidize. Reactions with some materials such as alkenes, ether and other compounds are highly unstable and explosive.				
Hazardous Decomposition	None. Ozone decomposes rapidly to oxygen (O2).				
Conditions to Avoid	Do not concentrate to high levels (>17%/wt.). The decomposition of ozone at high concentrations can become explosive.				
Incompatibility	Avoid contact with materials that can oxidize.				
Health Hazard Data					
Threshold Limit Value	The American Conference of Governmental Industrial Hygienists has set a threshold limit value for occupational exposure to ozone of 0.1 ppm as a time-weighted average over an 8-hour day. The short-term exposure limit is currently 0.3 ppm.				
Primary Route of Entry	Pulmonary system				
Effects of Single Overexposure	May cause irritation of the respiratory tract experienced as nasal discomfort, dryness, irritation of the throat, pain or congestion of the chest, difficult breathing or coughing. Irritation of the eyes, headache, nausea and drowsiness may also occur. Concentrations above 9 ppm have been found to result in pneumonia. Exposure to high concentrations could be fatal.				
Emergency First Aid					
Exposure	Symptom/Prevention	First Aid			
Emergency Overview	Ensure adequate ventilation in the area where ozone is present	Remove from the presence of air containing ozone.			
Inhalation	Irritating to respiratory system. Cough, headache, shortness of breath. Ventilation.	Remove from the presence of air containing ozone. Administer oxygen if necessary. If breathing is difficult or discomfort persists, obtain medical attention.			
Skin	Not an expected route of entry.				
Eyes	Irritating to eyes. Ventilation. Face shield or eye protection with breathing protection.	Remove from the presence of air containing ozone. Rinse with water for several minutes and seek medical attention if necessary.			
Ingestion	Not an expected route of entry				
	Exposure Control/Personal Protection				
Engineering Controls	Ozone equipment should be operated with an ozone off-gas destruct process.				
Ventilation	Ozone off gas should be collected and destroyed prior to release.				
Eyes/Face	Not applicable				
Skin	Not applicable				
Respiratory	Respirator or self-contained breathing apparatus for concentrations greater than 0.1ppm.				
Handling	Not applicable				
Storage	Ozone gas cannot be stored. Ambient ozone gas monitors should be used for detection.				
Disposal Information					
Waste Disposal		Ozone rapidly decomposes to form oxygen (O2). Use an ozone destruct system to convert any unused ozone or off gas into oxygen prior to discharge.			